

Transitioning of Spent AGR Fuel from Wet to Dry Storage

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Introduction

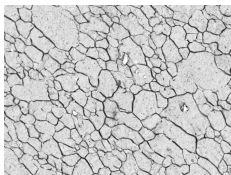
- Large quantities of nuclear fuel has accumulated in storage ponds.
- Fuel can not be stored in ponds indefinitely.
- In many countries dry storage of spent nuclear fuel (SNF) is used as an interim measure.
- Key requirements for dry storage are criticality prevention, integrity maintenance and retrievability.
- All of these are affected by corrosion and hence the interaction with water.



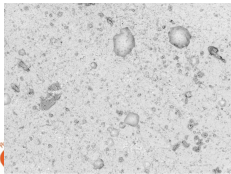
Corrosion Behaviour

- Differences in corrosion behaviour of inner and outer cladding surface.

Inner Surface



TM3030Plus0338 2016/05/13 11:40 I 100 µm

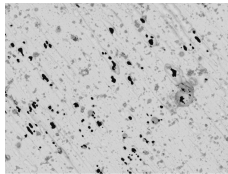


TM3030Plus0261 2016/03/17 08:56 I 100 µm

Outer Surface

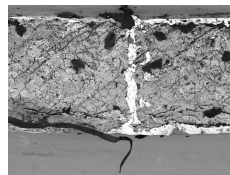


TM3030Plus0324 2016/05/13 11:11 I 100 µm

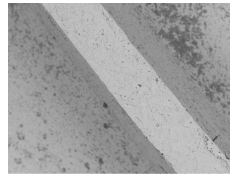


TM3030Plus0254 2016/03/17 08:37 I 100 µm

Rib



TM3030Plus0325 2016/05/13 11:14 I 200 µm

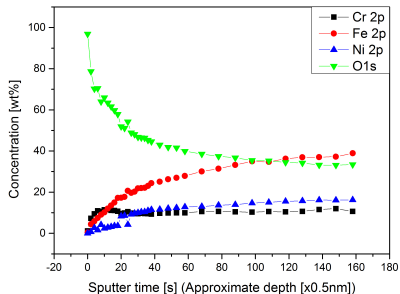


TM3030Plus0256 2016/03/17 08:41 I 500 µm

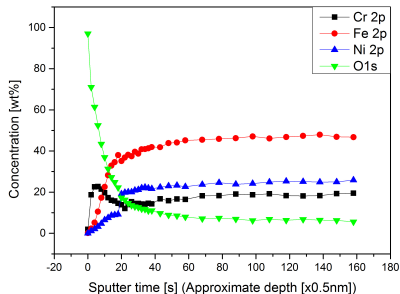
XPS Depth Profiling

- Outer surface appears to have lower oxygen content and thinner oxide layer.

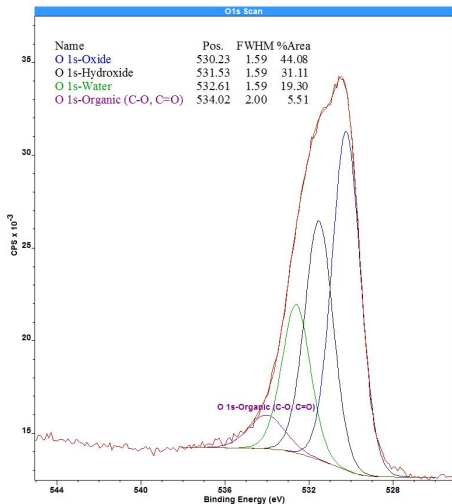
Inner



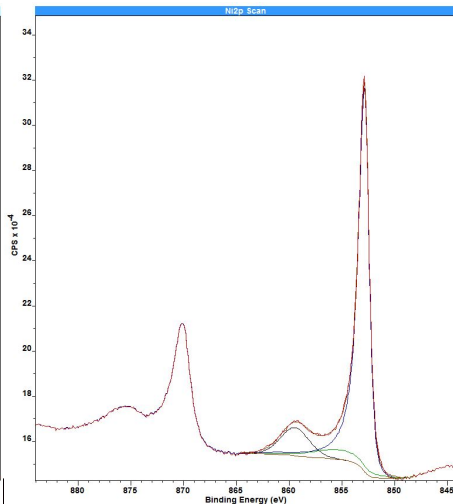
Outer



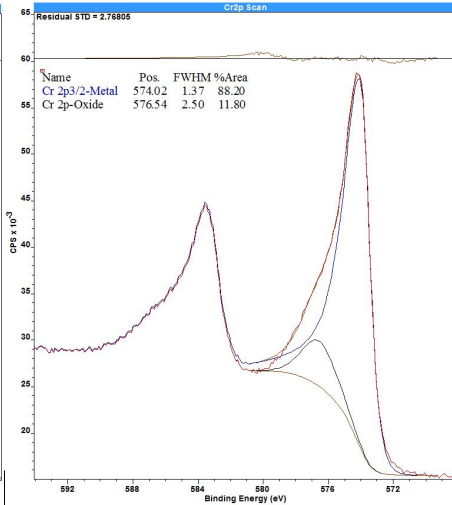
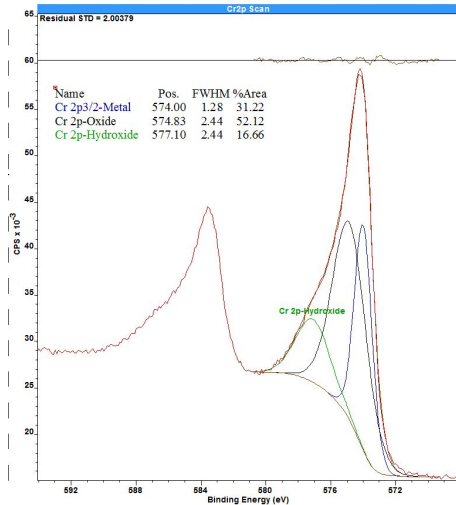
Oxygen



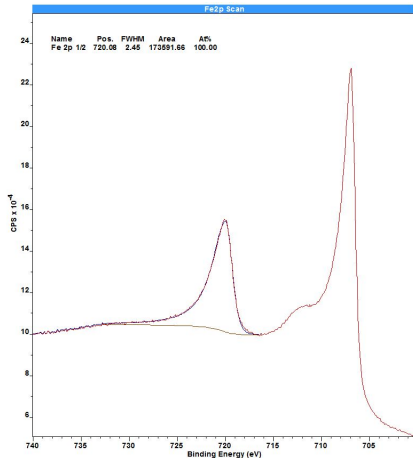
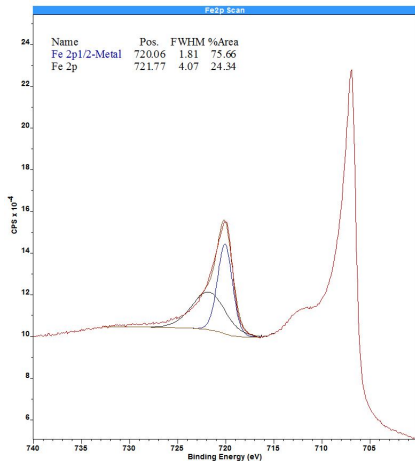
Nickel



Chromium. Expected peaks 574.3, 576 and 577.3 eV for metal, oxide and OH respectively.



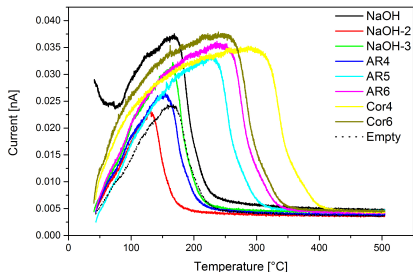
Iron. 3/2 peak overlaps with Ni LMM
Auger.



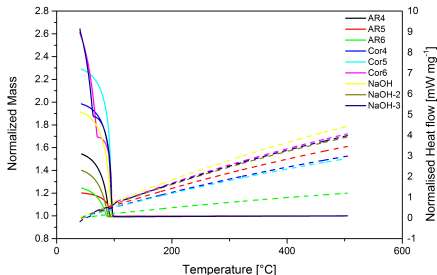
Adsorbed Water

- TGA/MS Shows insignificant adsorbed water.

MS

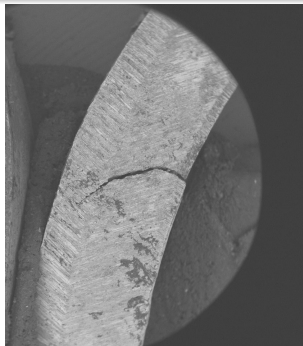


TGA/DSC



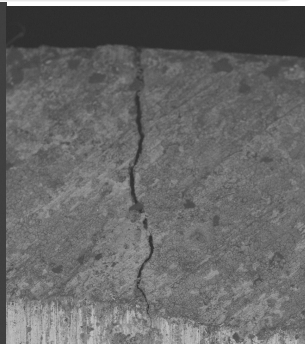
Cracked Samples

- Steel treated to form cracks.
- No success with 20/25 but have cracked 304.



2016/11/10 11:10 |

4 mm



2016/11/10 11:52 |

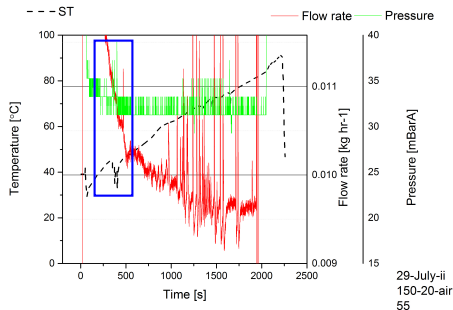
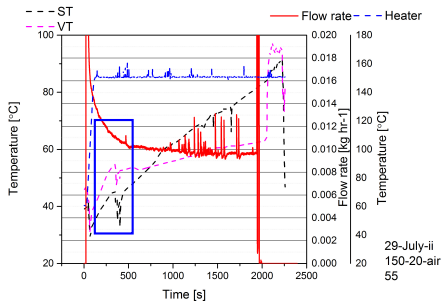
500 μm

Drying Rig and Test Piece



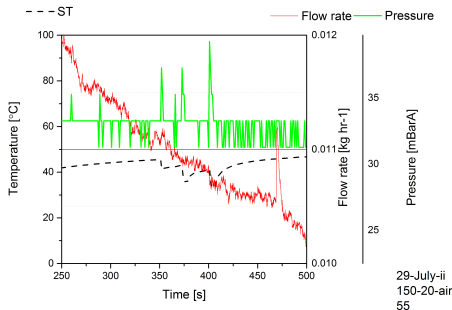
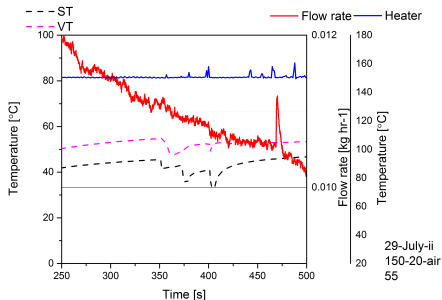
Drying Rig Data

Data output is something like this.



Drying Rig Data

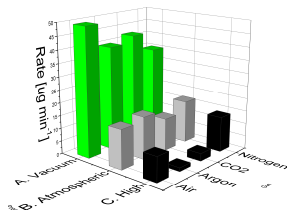
Zooming in on highlighted areas.



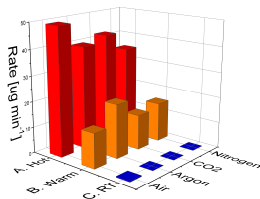
36 Different Conditions Used

- Three Temperatures; RT, 60°C (wam) and 150°C (hot).
- Three Pressures; 20 mBarA, atmospheric pressure, high pressure (10 BarG).
- Four different gases; air, N₂, Ar and CO₂.

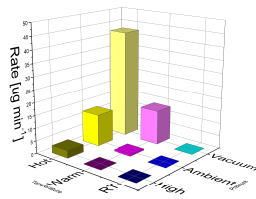
Hot drying



Vacuum Drying



Single gas-CO₂



Summary

Sample Preparation

- Adsorbed water is negligible.
- Still working on XPS and SCC.

Drying Tests

- Initial work has shown hot vacuum drying to be most effective.
- Further tests ongoing.

Grain Boundary Damage Mechanisms in strained AGR under Irradiation

C. Barcellini^{*1,2}, S. Dumbill³, S. Pimbott^{1,4}, and E. Jimenez-Melero^{1,2}

¹DCF, UoM, ²School of Materials, UoM, ³NNL, UoM, ⁴School of Chemistry, UoM

(*Correspondence: chiara.barcellini@postgrad.manchester.ac.uk)

DISTINCTIVE Theme 1 Meeting: AGR, Magnox and Exotic Spent Fuels

The Rheged Centre, Redhills, Penrith, 15th November 2016



1 Introduction

2 Materials and Methods

3 Experimental Results

4 Conclusions

5 Future Work



1	Introduction
2	Materials and Methods
3	Experimental Results
4	Conclusions
5	Future Work

Interim Storage of AGR Spent Fuel

- ▶ Wet storage of AGR fuel might last several years
 - ▶ challenge for the corrosion properties of the cladding material
- ▶ Through-wall failures during wet storage are ascribed to inter-granular corrosion (IGA)
- ▶ Lack of recent data about irradiated AGR cladding material
- ▶ Difficulties in studying neutron-irradiated materials



Sensitisation with charge particles

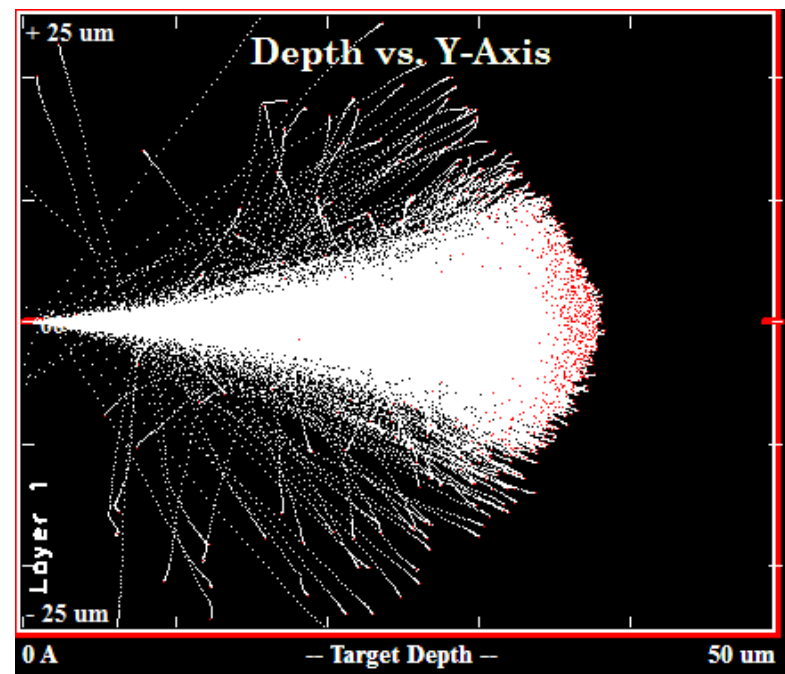
Sensitised specimens can be produced using intense beams of protons or heavy ions

ADVANTAGES:

- ▶ user-adjustable radiation
- ▶ similar microstructure
- ▶ reduced induced radioactivity

PhD main Objectives:

- ▶ access irradiation conditions
- ▶ study microstructures produced during irradiation



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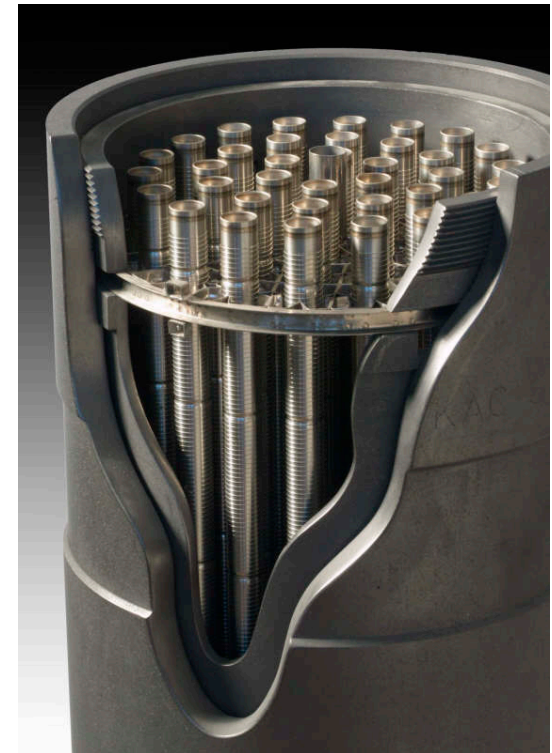
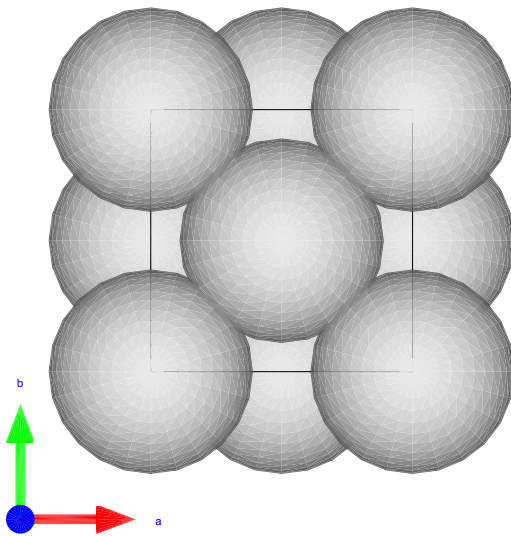
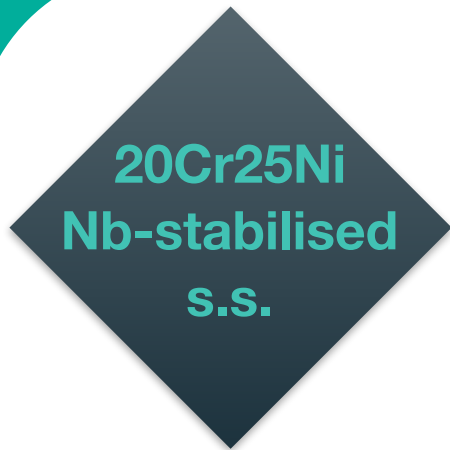


AGR cladding material

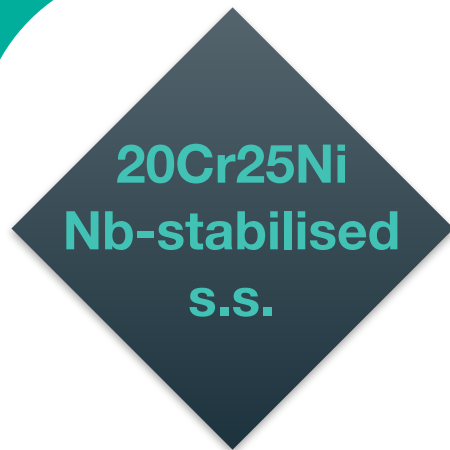
Austenitic Stainless Steel Fe-Ni-Cr alloy

- ▶ high corrosion resistance
- ▶ no ductile-brittle transition
- ▶ high creep resistance

Austenitic matrix has an
FCC crystal structure



Specimen Preparation



- cold work
- 40min at 930°C

Annealed
specimen

▸ irradiation

Irradiated
specimen

- austenitic matrix
 - composition -> SEM
 - g.b -> SEM & EBSD & TEM
 - texture -> EBSD
- second phases
 - composition -> SEM & TEM
 - texture -> EBSD

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 - composition -> SEM
 - g.b -> SEM & EBSD & TEM
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Specimen Preparation

**20Cr25Ni
Nb-stabilised
S.S.**

- cold work
- 40min at 930°C

**Annealed
specimen**

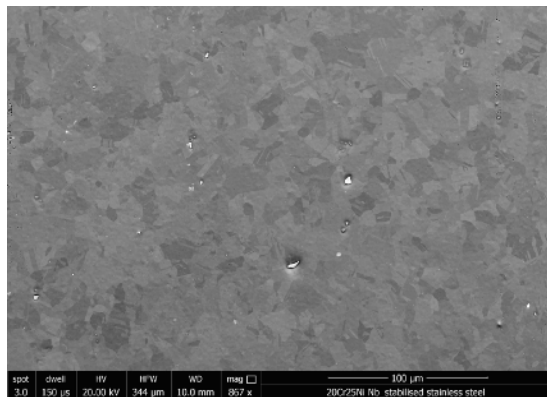
▸ irradiation

**Irradiated
specimen**

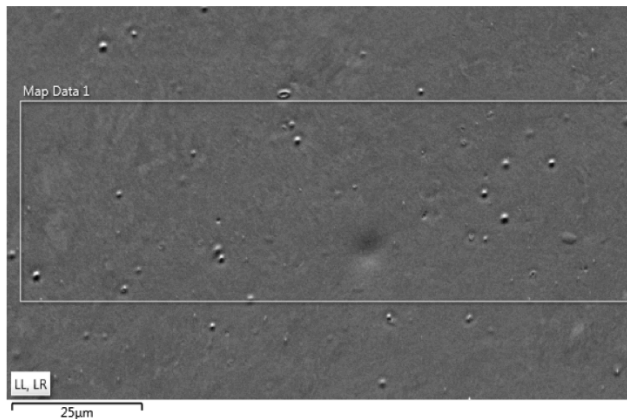
- austenitic matrix
 - composition -> SEM
 - g.b -> SEM & EBSD & TEM
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- second phases
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- austenitic matrix
 - composition -> SEM
 - g.b -> SEM & EBSD & TEM
 - texture -> EBSD
- second phases
 - composition -> SEM & TEM
 - texture -> EBSD

BATCH 1
930°C for 40min



BATCH 2

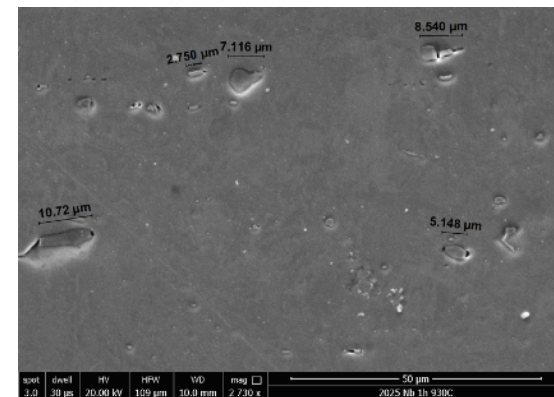


Materials as received



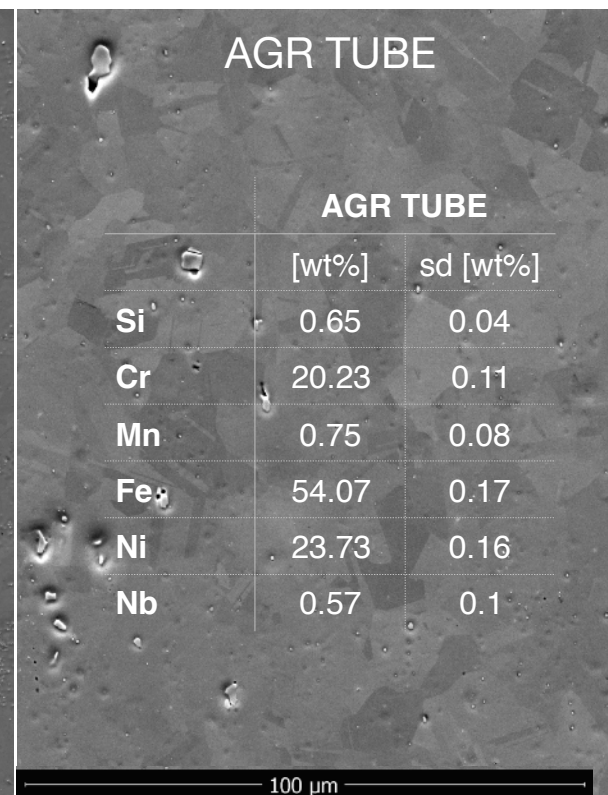
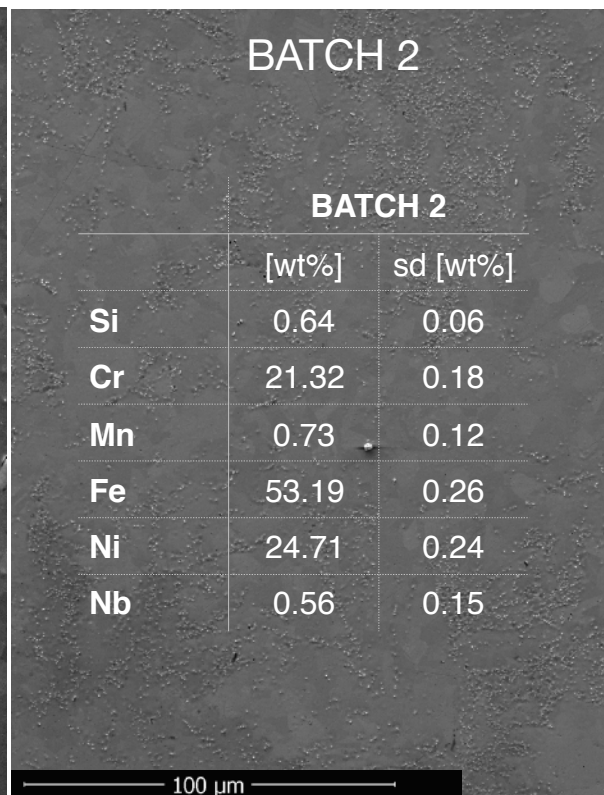
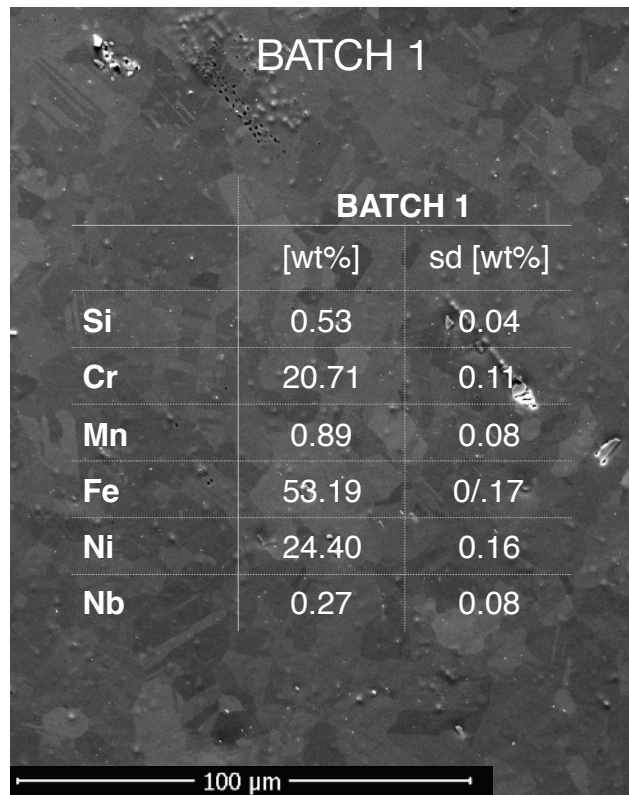
AGR TUBE

- cut
- outer ribs removed (0.5mm thick)
- split
- rolled until flat

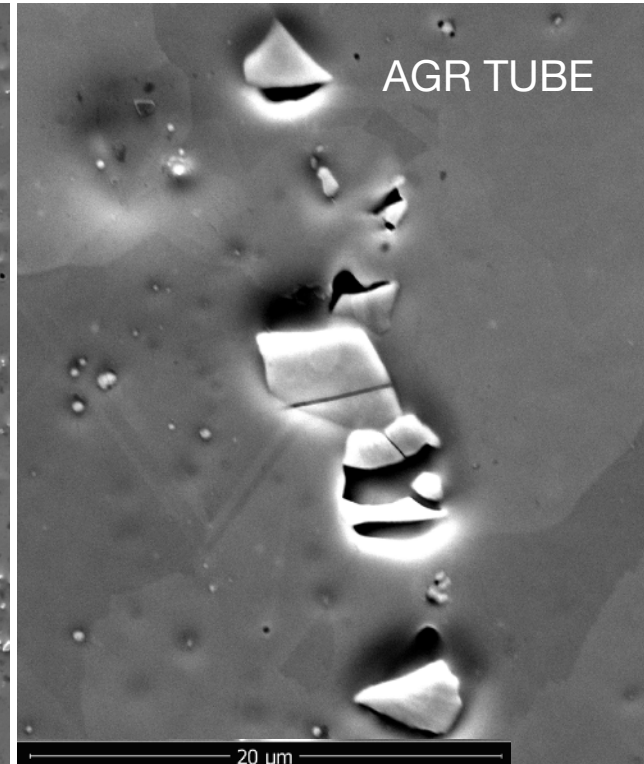
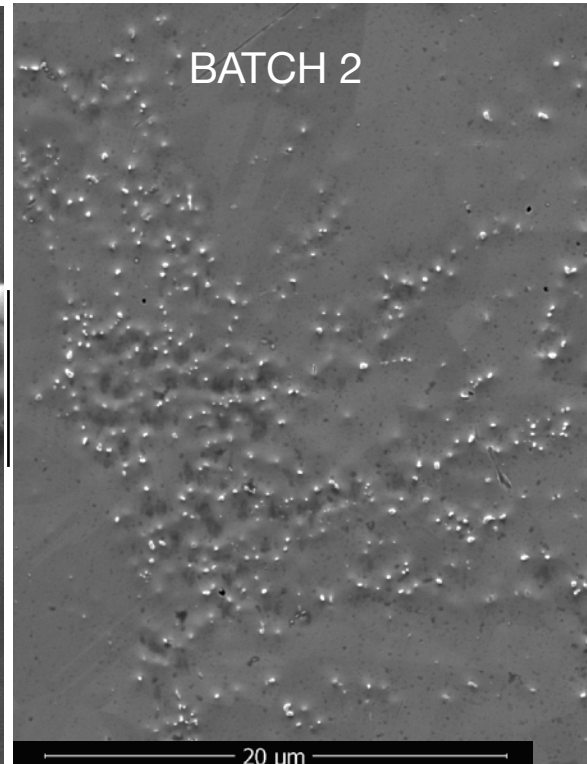
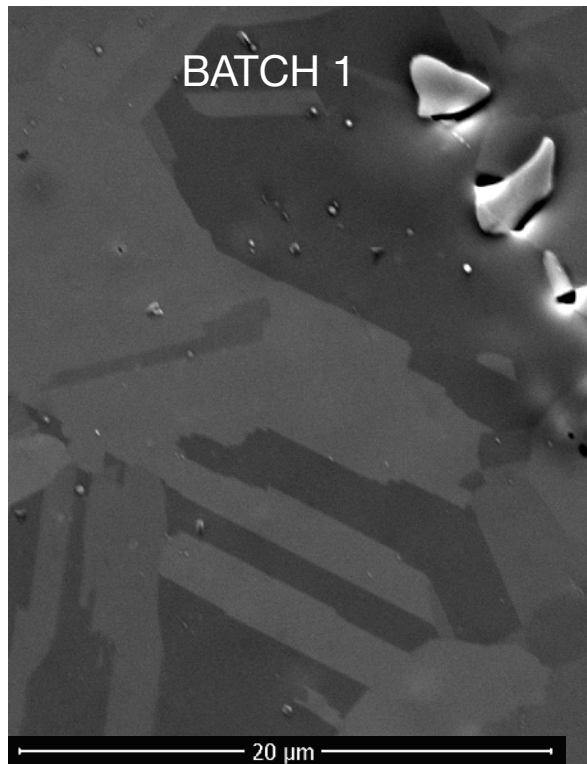


- heat treat 930°C for 40min

Annealed 20Cr15N Nb-stabilised s.s.

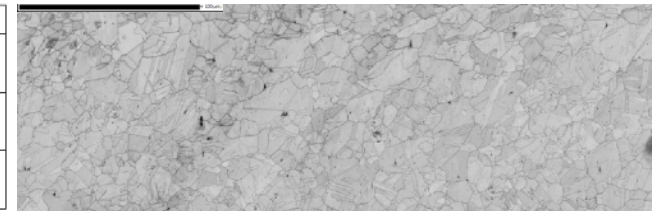
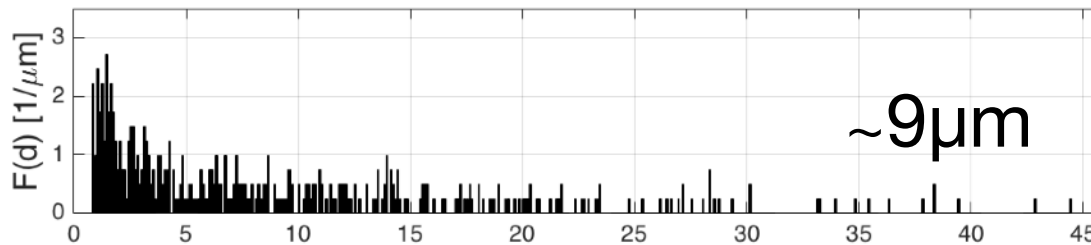


Annealed 20Cr15N Nb-stabilised s.s.

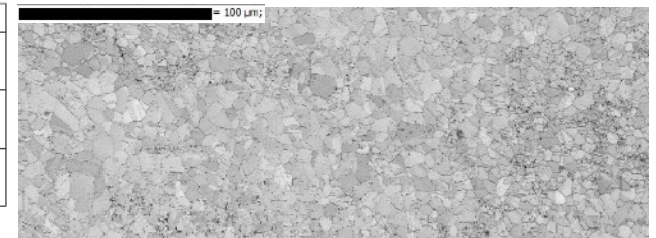
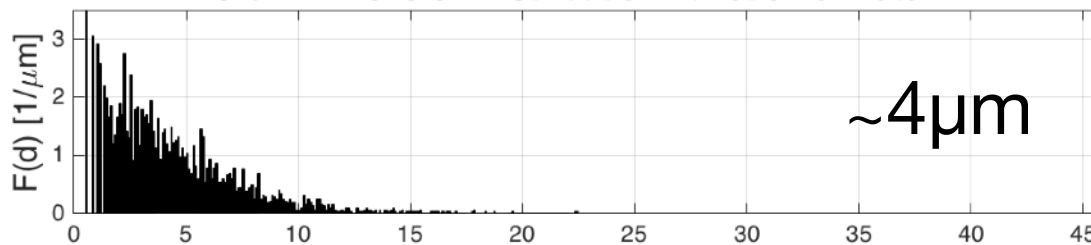


Recrystallisation

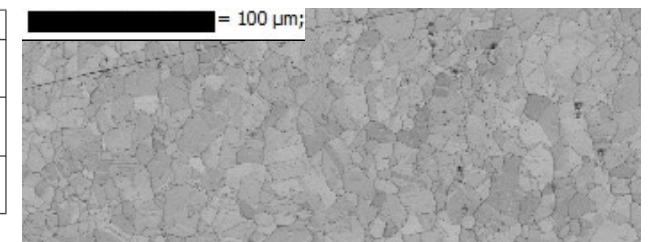
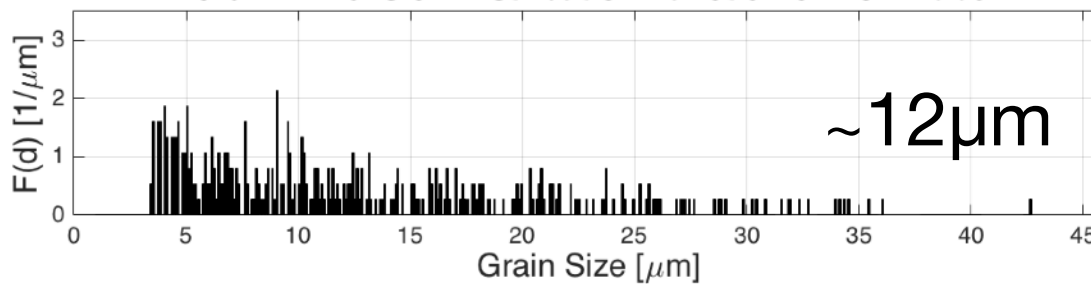
Grain Dimension Distribution Function of Batch 1

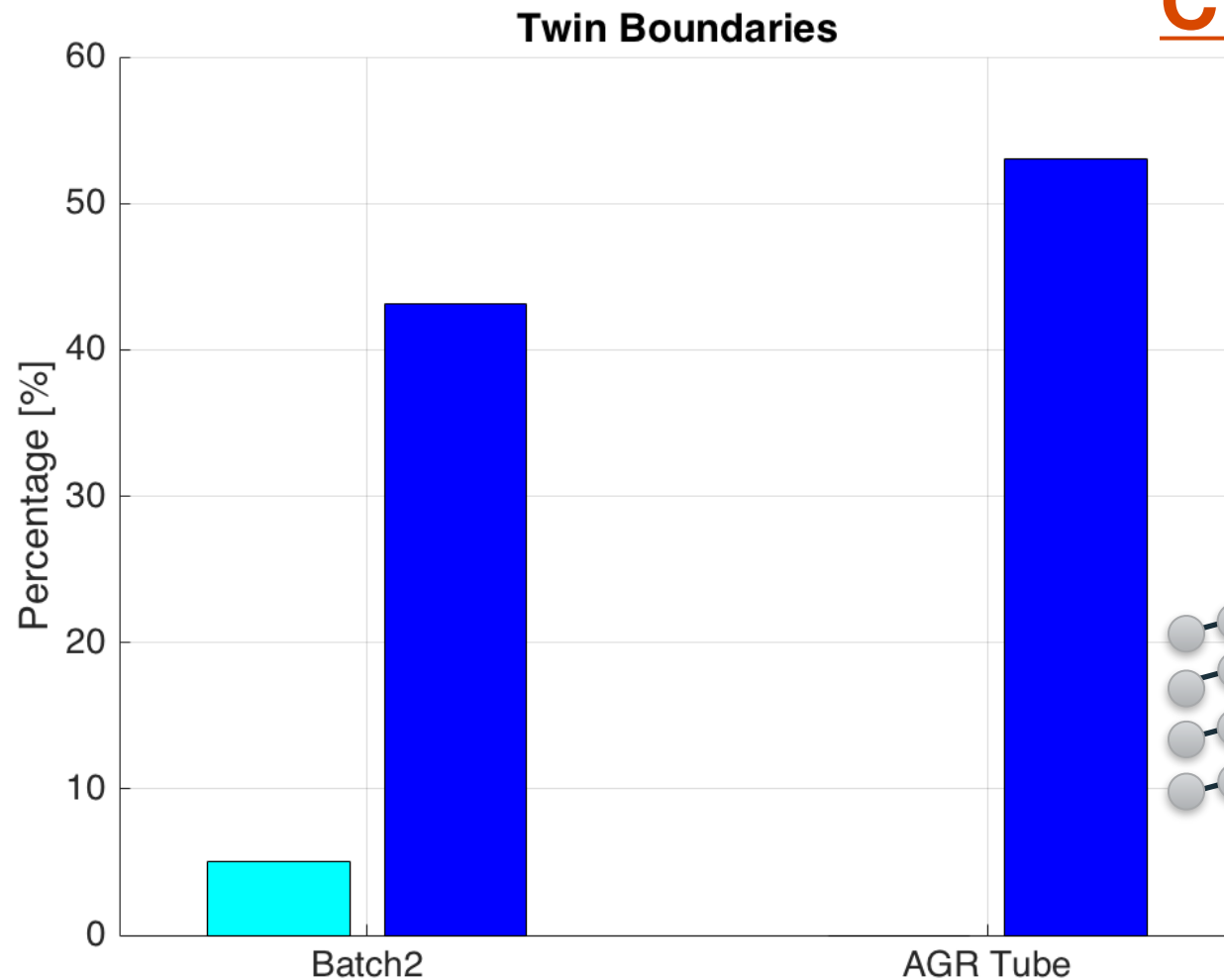


Grain Dimension Distribution Function of Batch 2

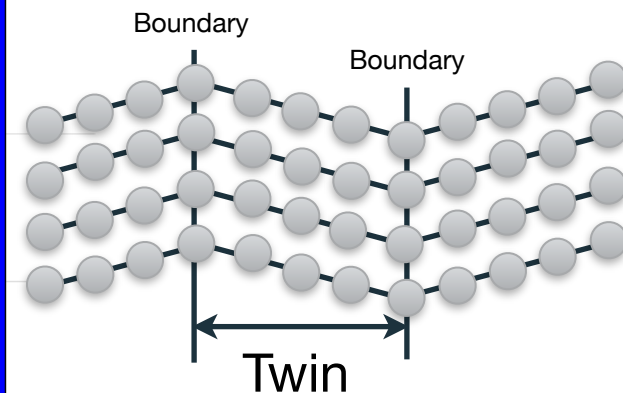
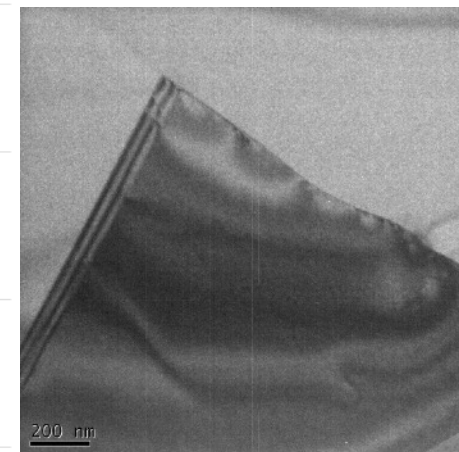


Grain Dimension Distribution Function of AGR Tube

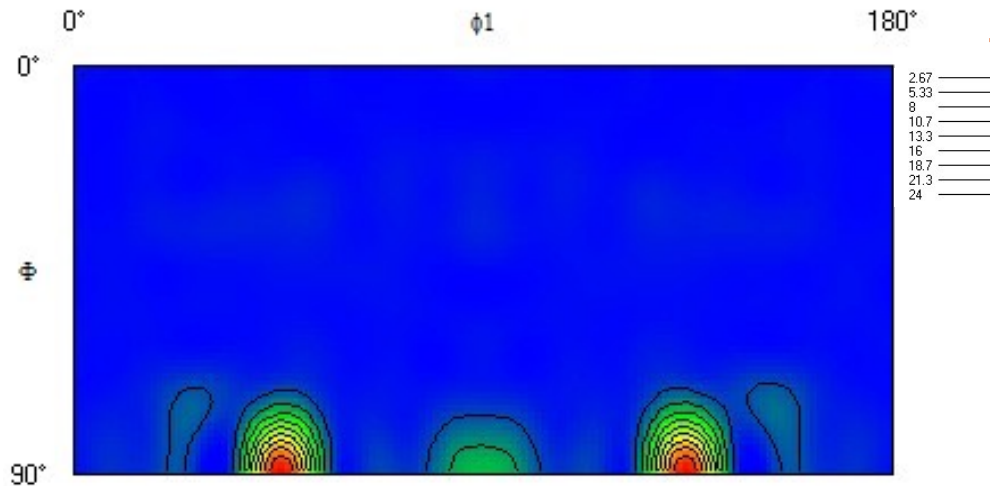




CLS Boundaries



Texture Development

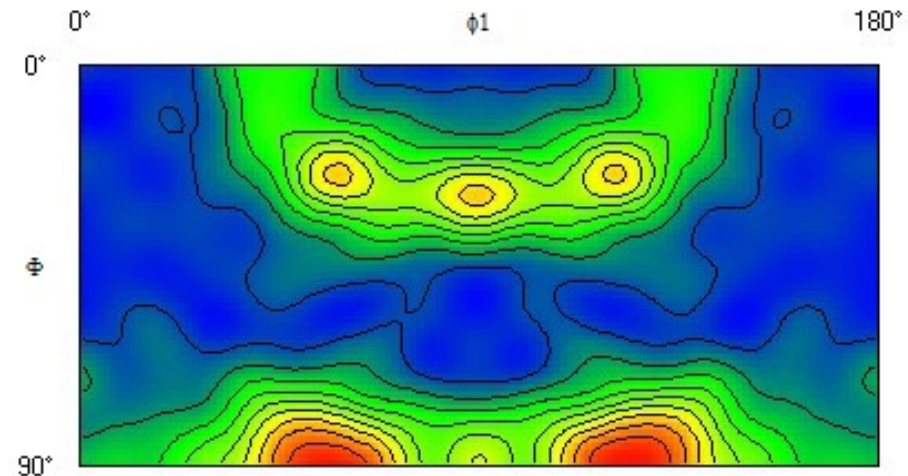
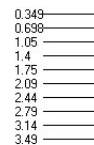


BATCH 2
as received

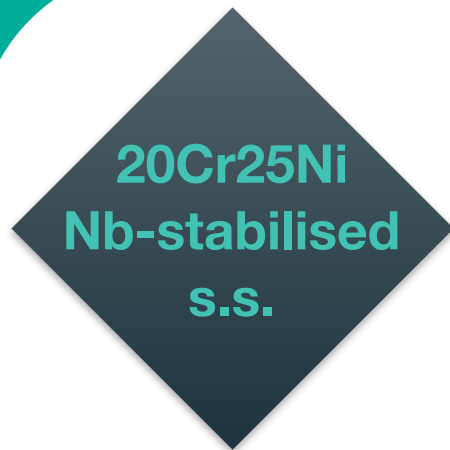
Goss $\{011\}\langle 100 \rangle$
Brass $\{110\}\langle 112 \rangle$

BATCH 2
1h at 930°C

Goss $\{011\}\langle 100 \rangle$
Brass $\{110\}\langle 112 \rangle$
Cube $\{001\}\langle 100 \rangle$
Copper $\{110\}\langle 111 \rangle$



Specimen Preparation



- cold work
- 40min at 930°C



- irradiation



- austenitic matrix
 - composition -> SEM
 - g.b -> SEM & EBSD & TEM
 - texture -> EBSD
- second phases
 - composition -> SEM & TEM
 - texture -> EBSD

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 - composition -> SEM
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 - texture -> EBSD
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 - composition -> SEM & TEM
 - texture -> EBSD



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Conclusions

- ▶ Proton irradiations of 20Cr25Ni Nb-stabilised s.s. can be performed in controlled conditions of damage and temperatures at DCF
- ▶ The heat treatment (40min at 930°C) produces a partially recrystallised austenitic structure with intra and inter-granular Nb-rich second phases
- ▶ The annealed austenitic matrix has a $\{101\}\langle 111 \rangle$ texture

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Future Work

- ▶ Micro-structural characterisation of heat treated samples
- ▶ New irradiation experiment in January 2017
- ▶ Preparation of irradiated specimens for AEM using
 - ▶ electropolishing
 - ▶ (FIB) - > Site-specific specimen preparation
 - ▶ grain boundaries -> RIS
 - ▶ second phases -> precipitates stability under irradiation

Thank you for your attention





Questions?



Bibliography

- [1] C. Taylor, Radiation-induced sensitisation of stainless steel, Berkeley Nuclear Laboratories, 1986.
- [2] D.I.R. Norris, C. Baker, J.M. Titchmarsh, Radiation-induced sensitisation of stainless steel, Berkeley Nuclear Laboratories, 1986.
- [3] P. T. Wady, A. Draude, S. M. Shubeita, A. D. Smith, N. Mason, S. M. Pimblott, E. Jimenez-Melero, Nucl. Instr. Meth. Phys. Res. A 806 (2016) 109.

Investigations into the formation of Uranium Hydride

James Ed Darnbrough

DISTINCTIVE Theme 1 meeting

Penrith

15/11/16

Talk Outline

- Interest
- Bulk investigation
- In-situ Synchrotron
- STEM and EELS study
- Conclusions

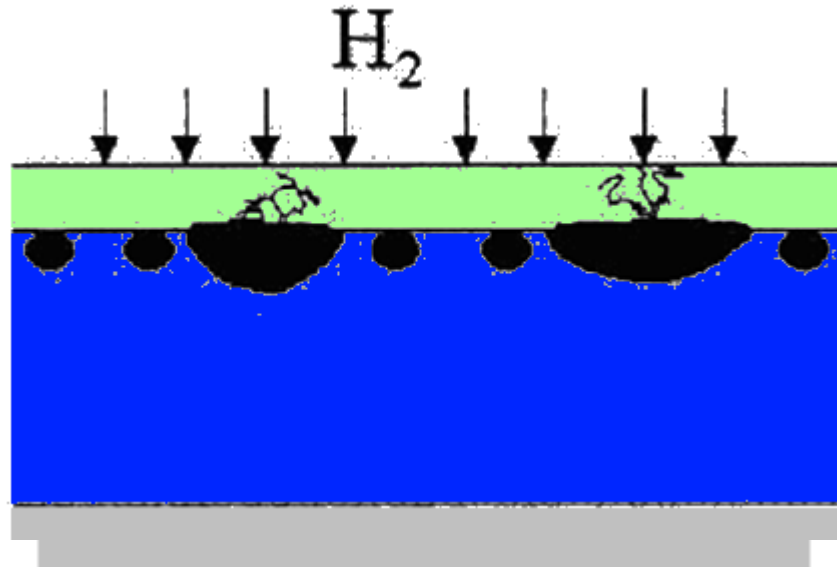
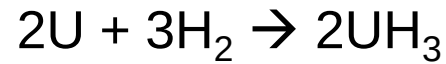
Interest in Uranium metal corrosion

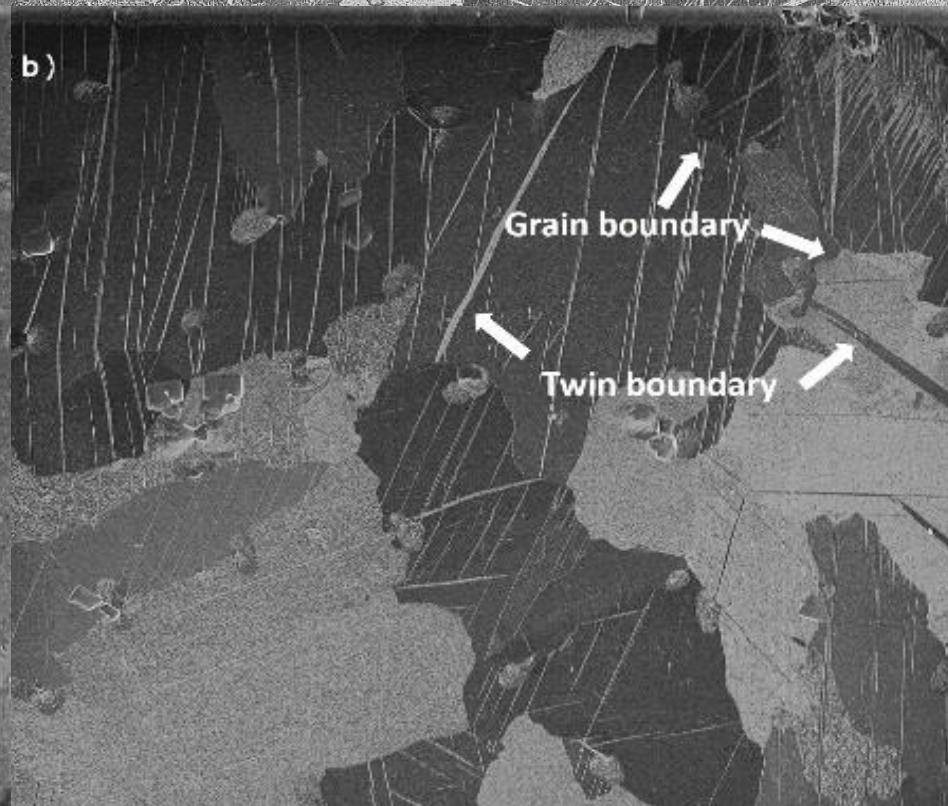
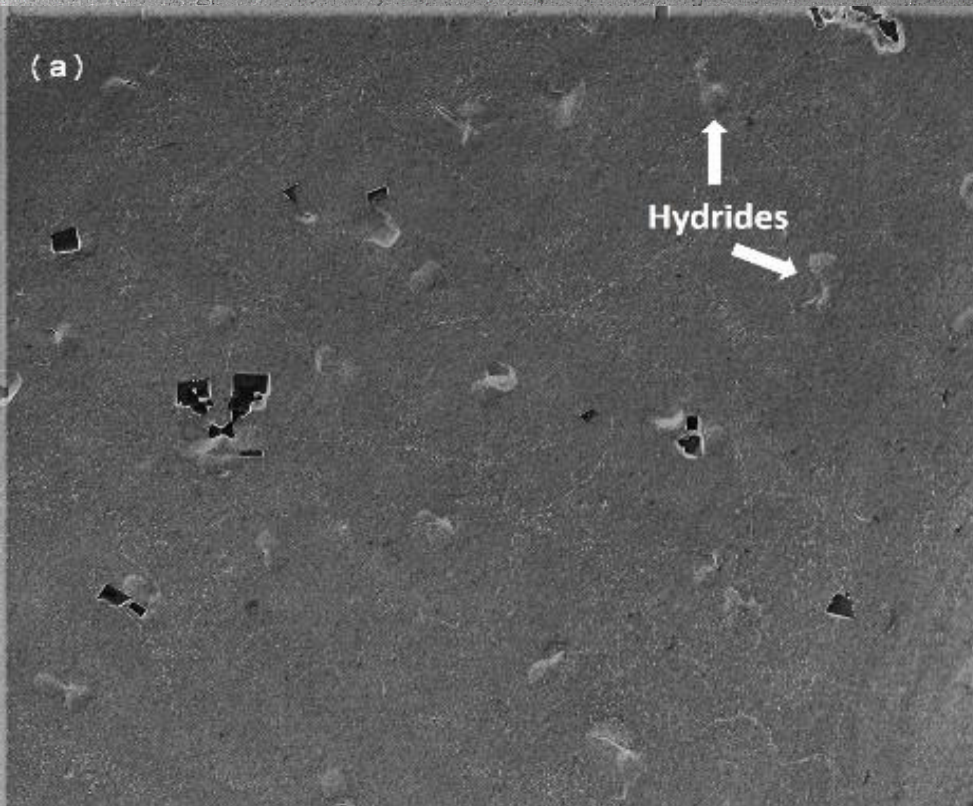
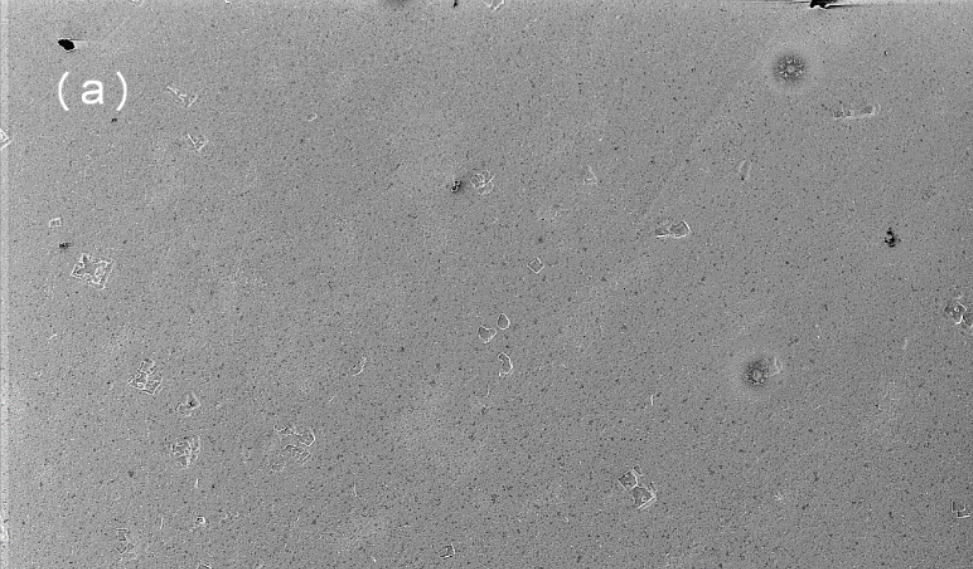
- Legacy uranium fuels
 - Magnox fuel (UK),
- High density metal fuels (Advanced Test Reactors)
 - Uranium Aluminium, Uranium Zirconium, Uranium Silicon, Uranium Molybdenum
- Analogous to Pu

Uranium Hydride



Proposed Hydride Formation





Beam	Tilt	Mag	Scan	pA	HFW
30.0 kV	0.4°	2.50 kX	H 45.26 s	69.0	122 μm

20 μm

Beam	Tilt	Mag	Scan	pA	HFW
30.0 kV	0.4°	2.50 kX	H 45.26 s	67.0	122 μm

20 μm

(c)

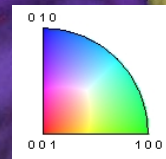
Blisters seen at
triple points

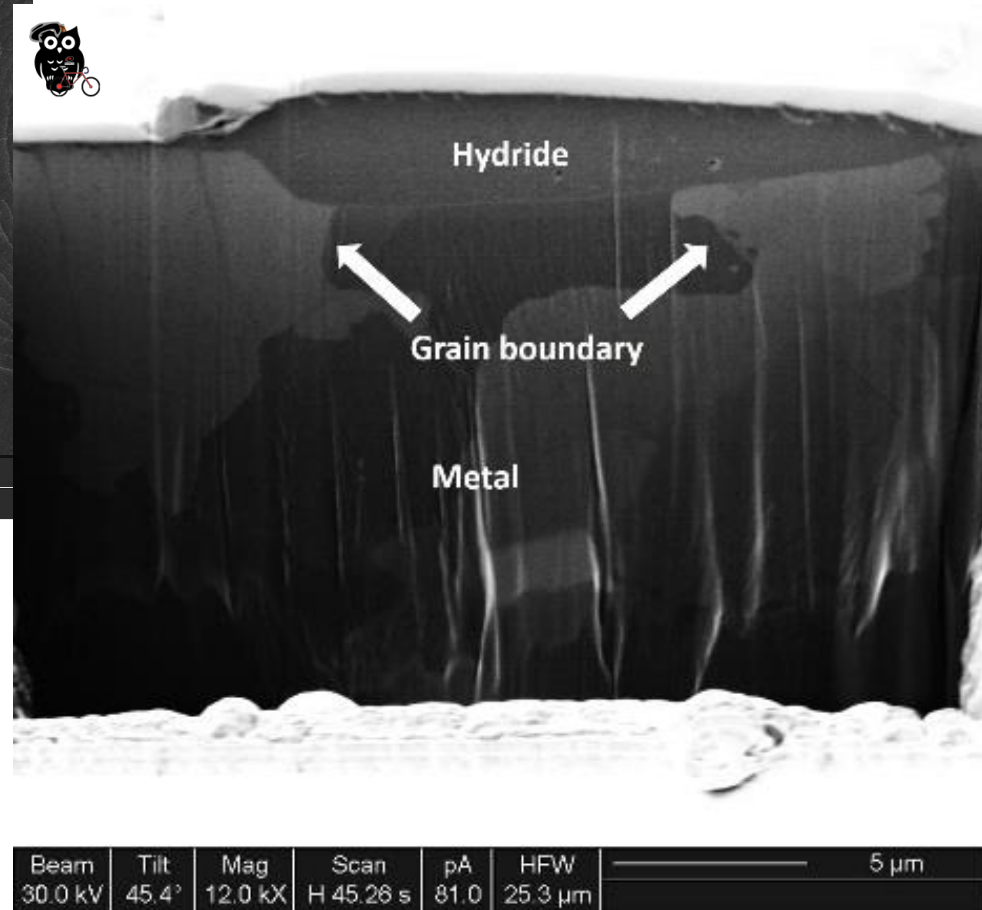
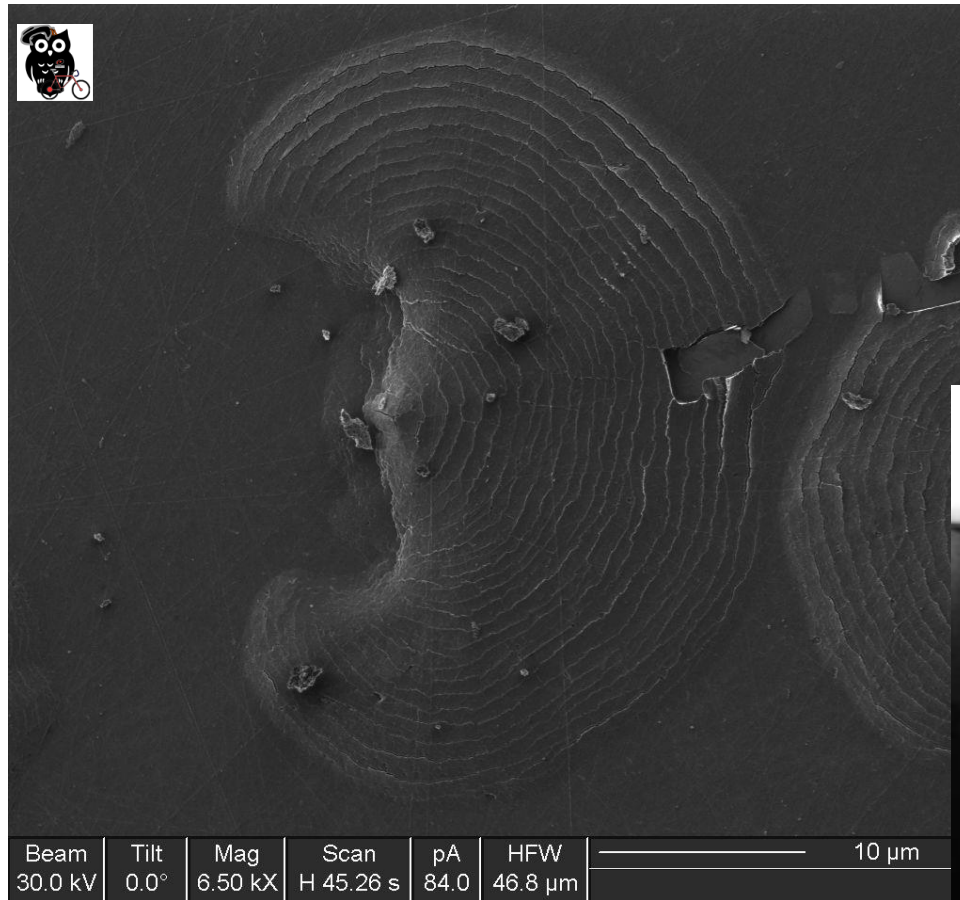
60 μm

	Min	Max
—	2°	15°
—	15°	65°
—	65°	180°

Blisters seen at high
angle boundaries

60 μm





U 19.1 g/cm³

↓
UH₃ 10.95 g/cm³

Significance of Blister Location

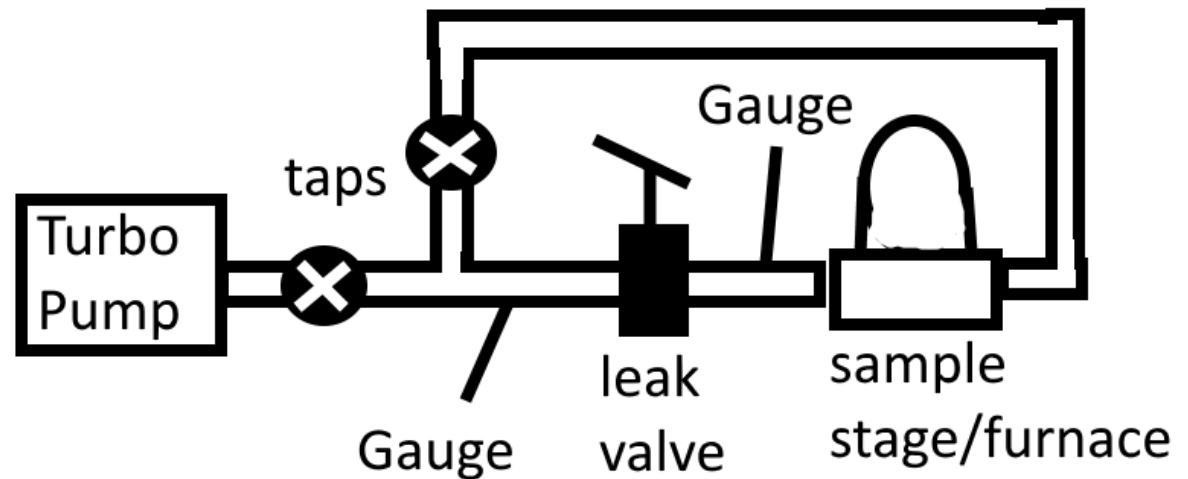
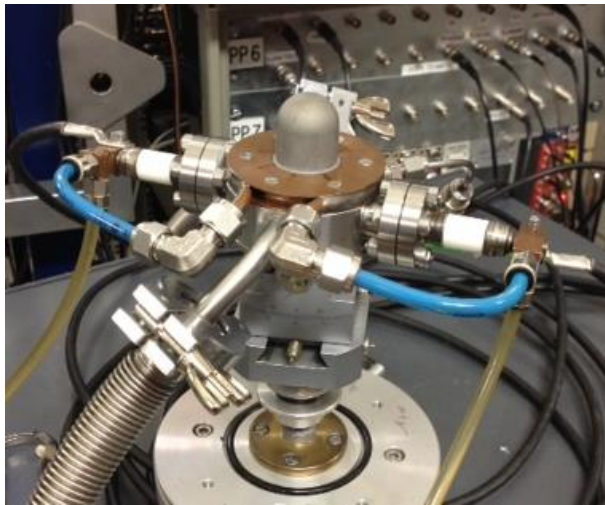
- At Triple points and high angle grain boundaries
- High diffusion paths?
- Free surfaces with lower density than crystalline material?
- Preferred growth direction? 
- Where does it start in uranium metal? 

in-situ experiment

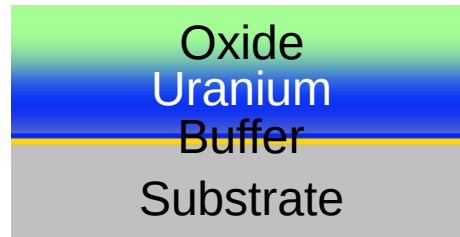
- Thin film samples produced (by DC magnetron sputtering) to mimic interface without large masses reducing the activity to exempt limits
- Samples are produce a uniform homogeneous oxide thickness on a known metal thickness
- Utilised high angle diffraction to look at crystallinity and low angle reflectivity to look at layer interfaces

Experimental set up

- Used XMaS beamline at ESRF
- Furnace mounted on goniometer allowed remote temperature control (RT-200°C) and gas dosing (1000mbar of 4% H_2/Ar)

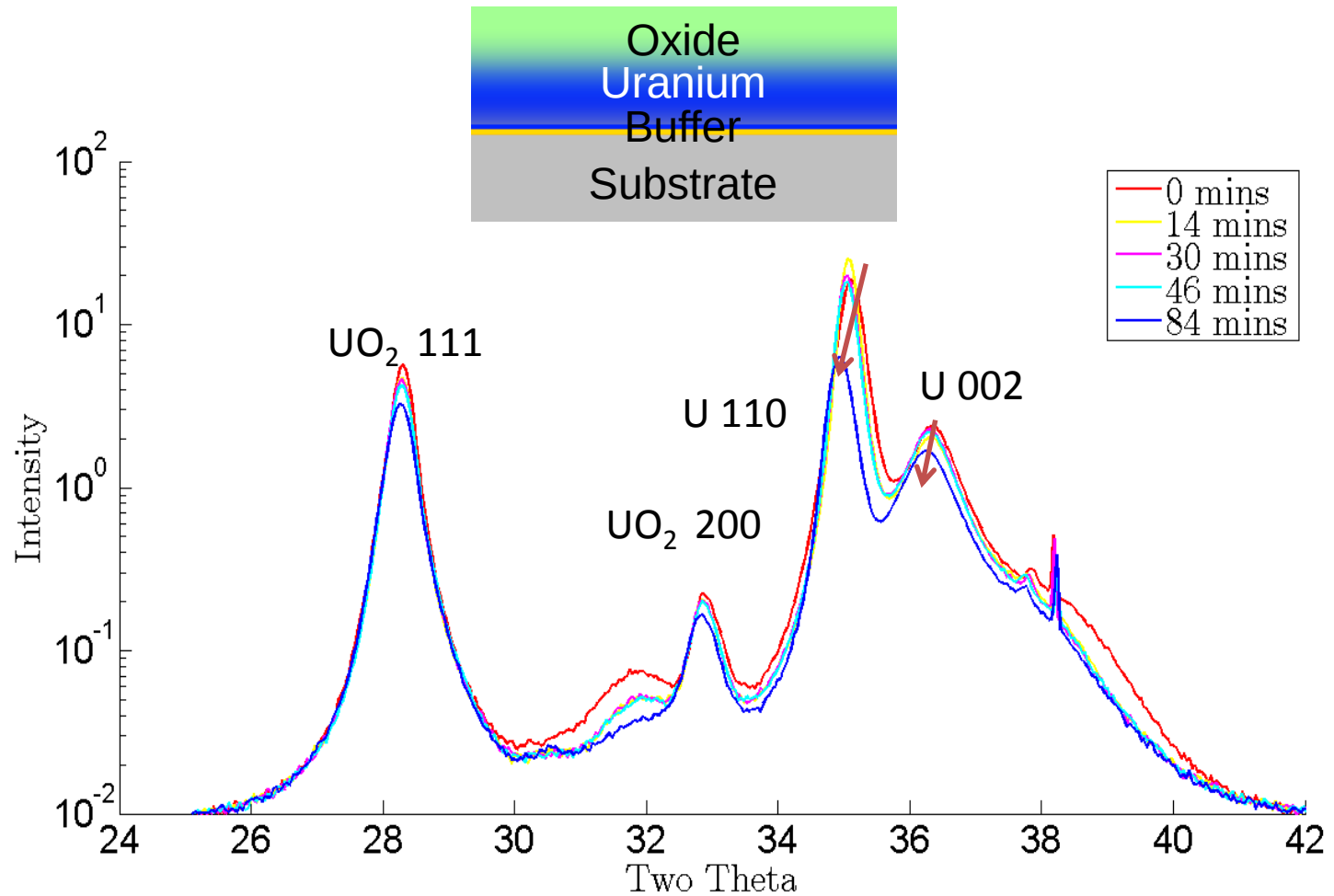


Introduction of H₂

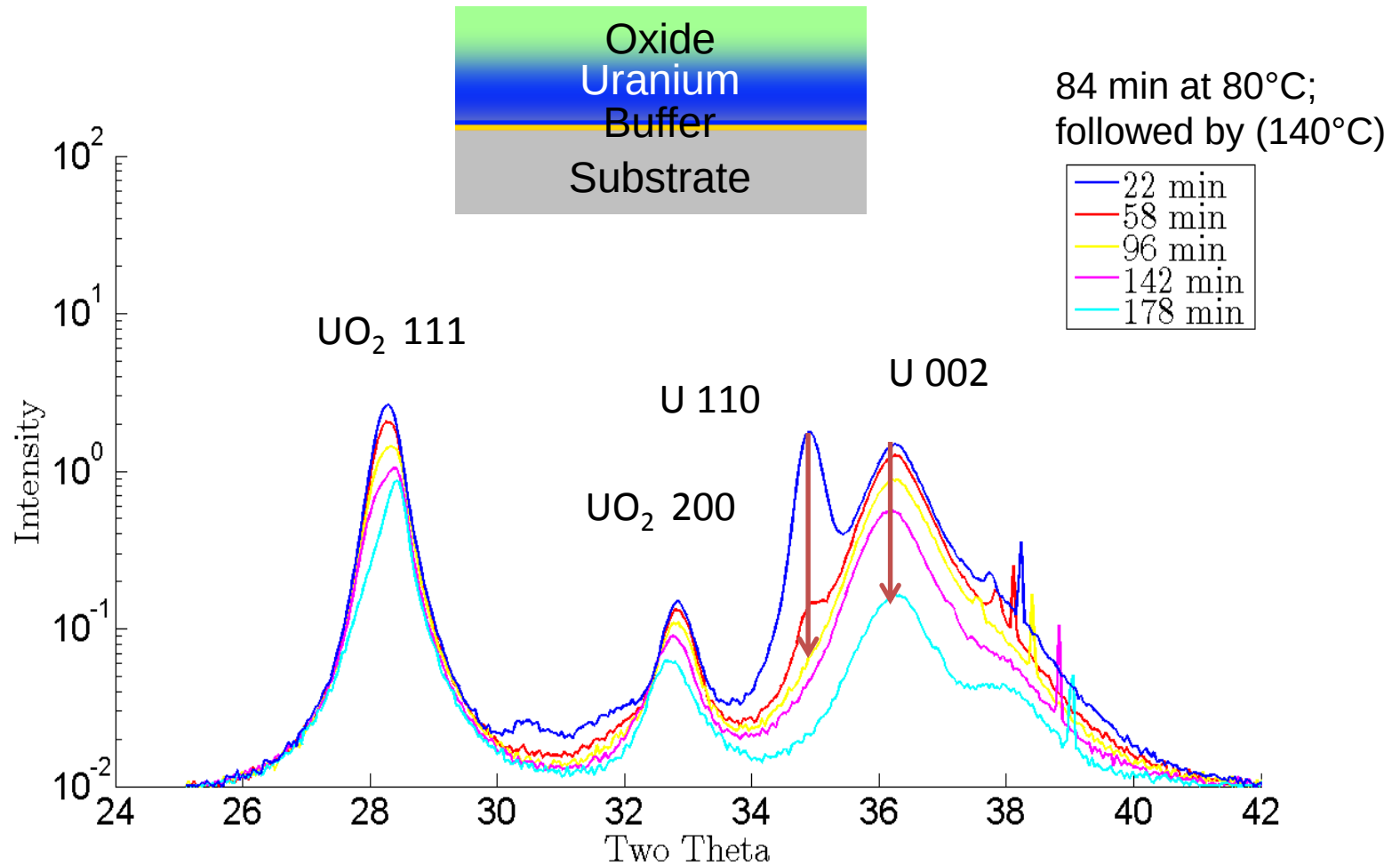


- U main orientation (110)
- U (002) intensity at approximately 10% of (110)
- UO₂ more intense predominantly (111) with some (200); higher UO_{2+x} oxides also observable
- Hydrogen Exposure
 - 84min 50 mbar gas mix (2 mbar H₂) at 80°C
 - Further 187min at 400 mbar gas mix (16 mbar H₂) at 140°C
 - And 156min 400 mbar gas mix (16 mbar H₂) at 200°C

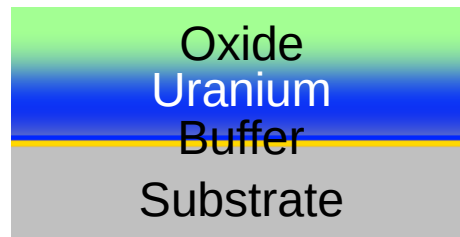
80 °C exposure



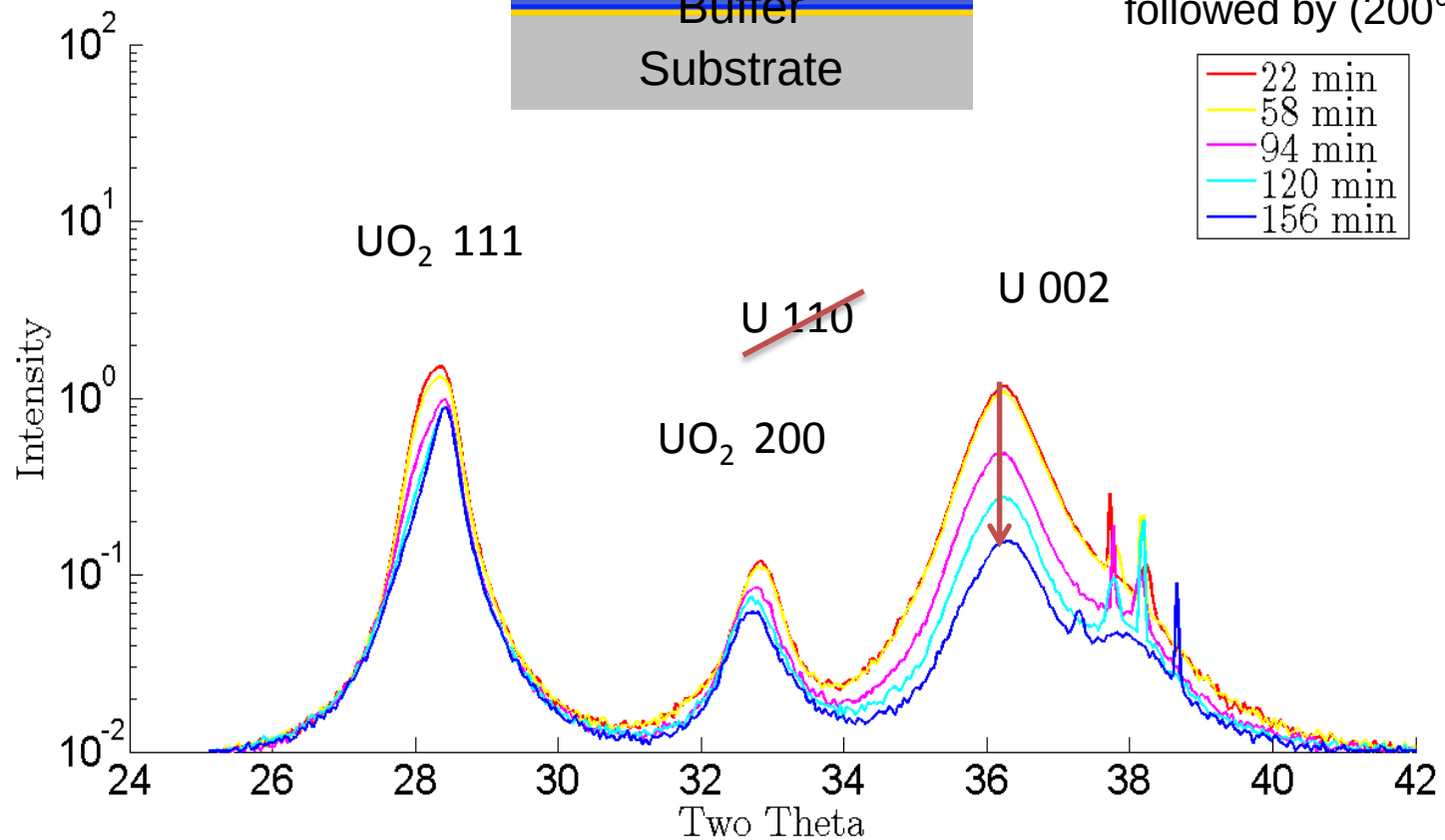
140 °C exposure



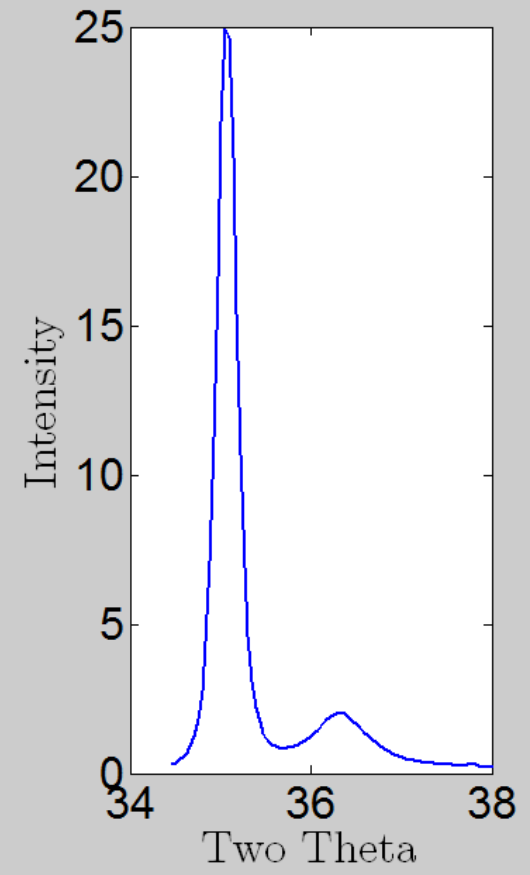
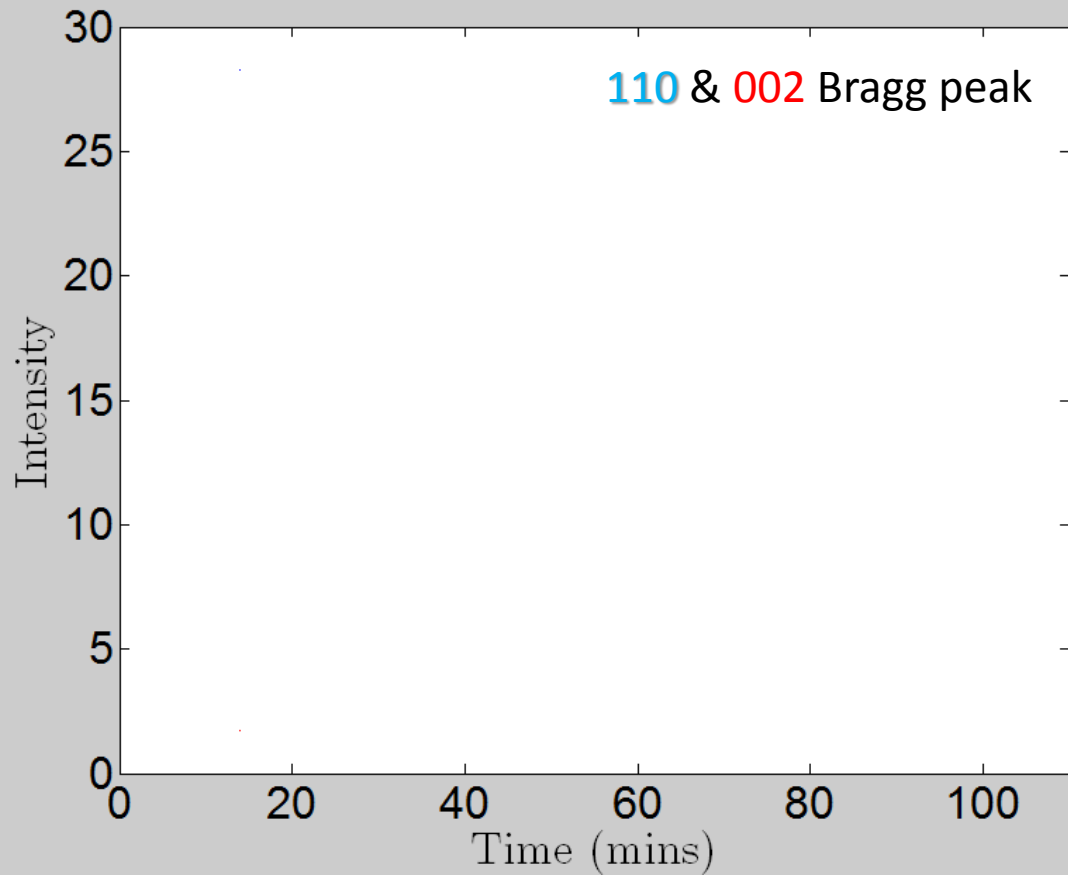
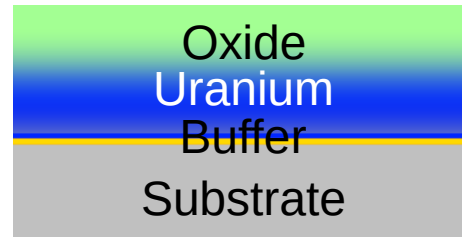
200 °C exposure



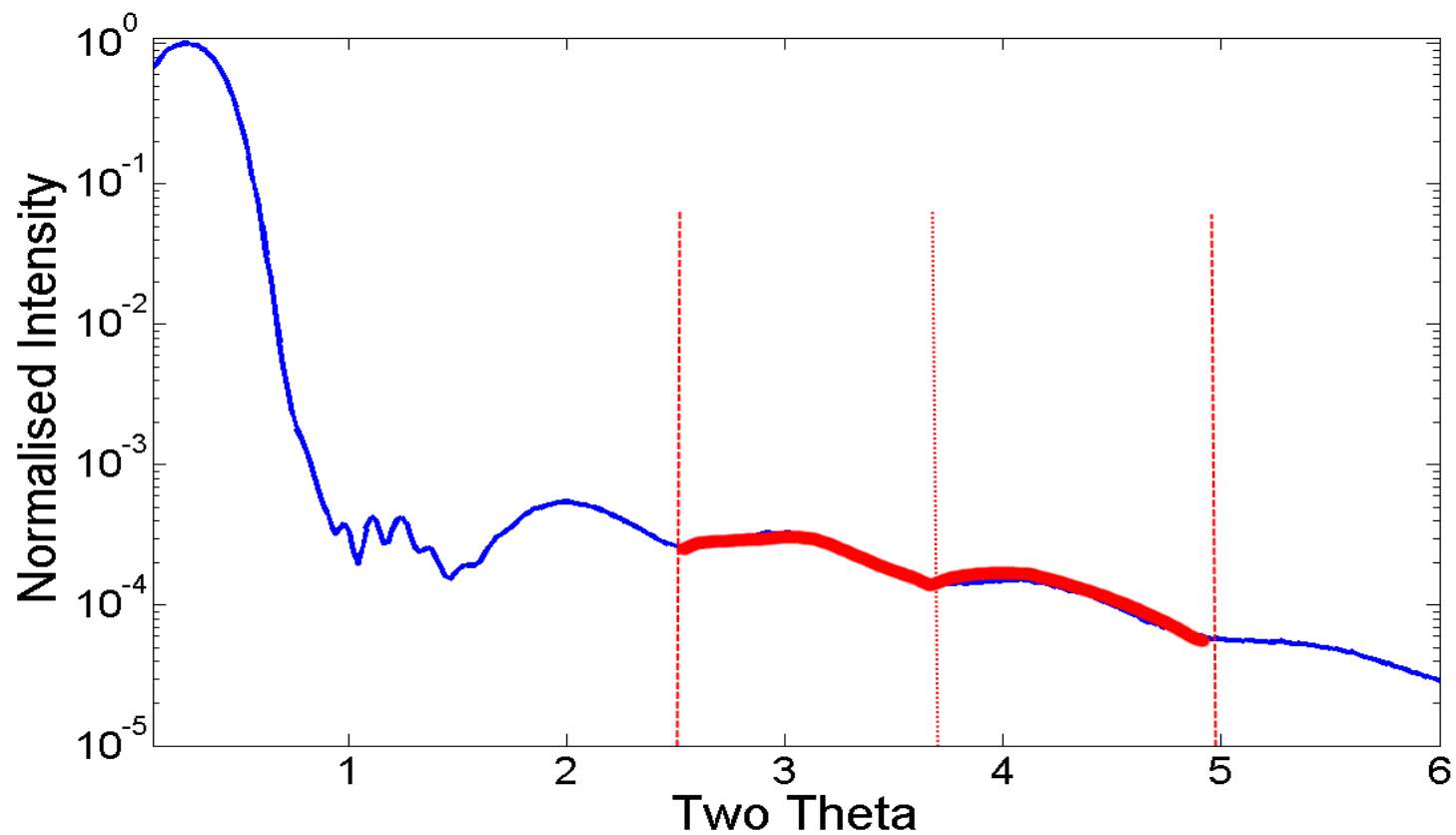
84 min at 80°C;
178 min at 140°C
followed by (200°C)



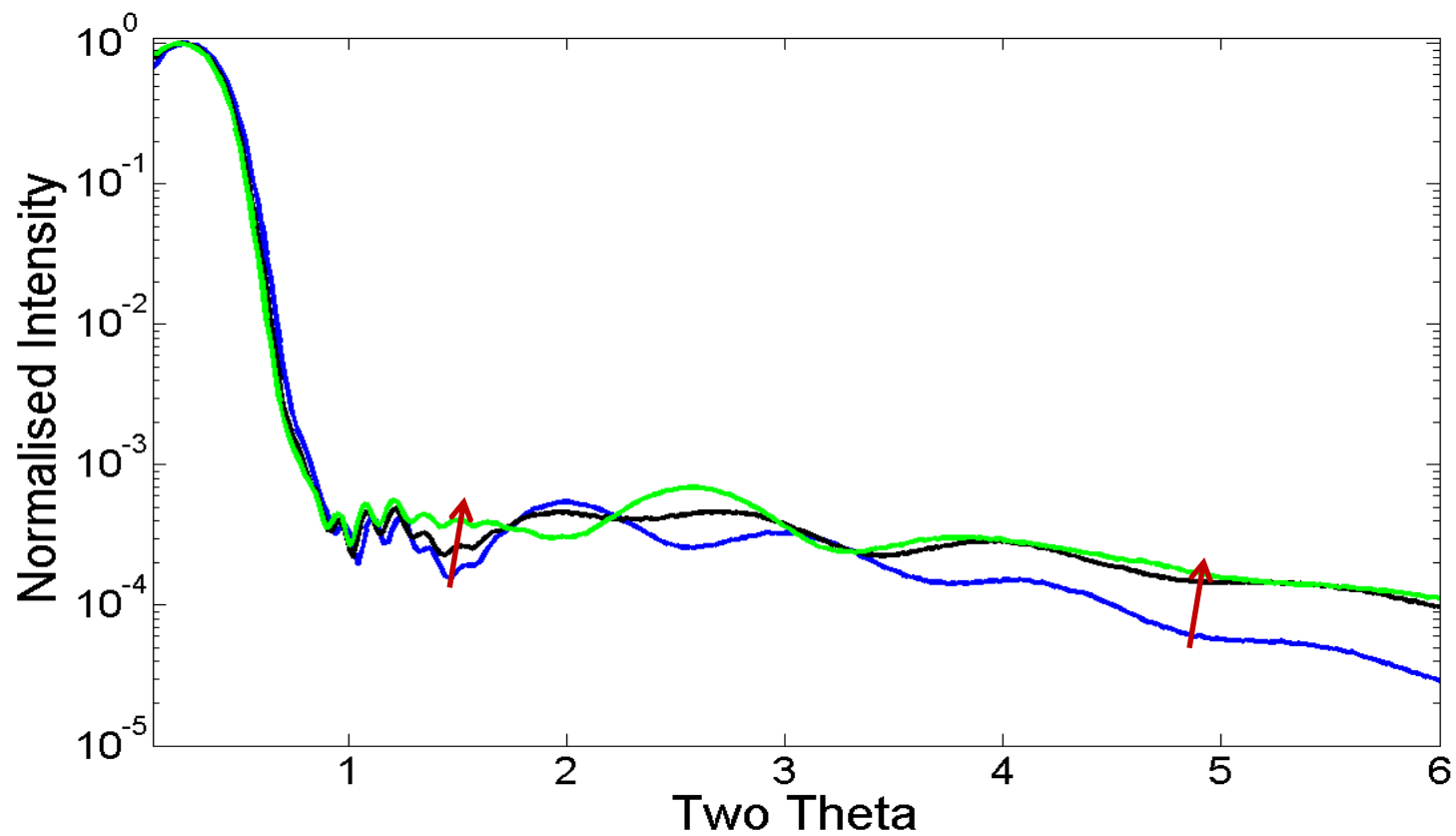
Rate of Crystalline Removal



Reflectivity



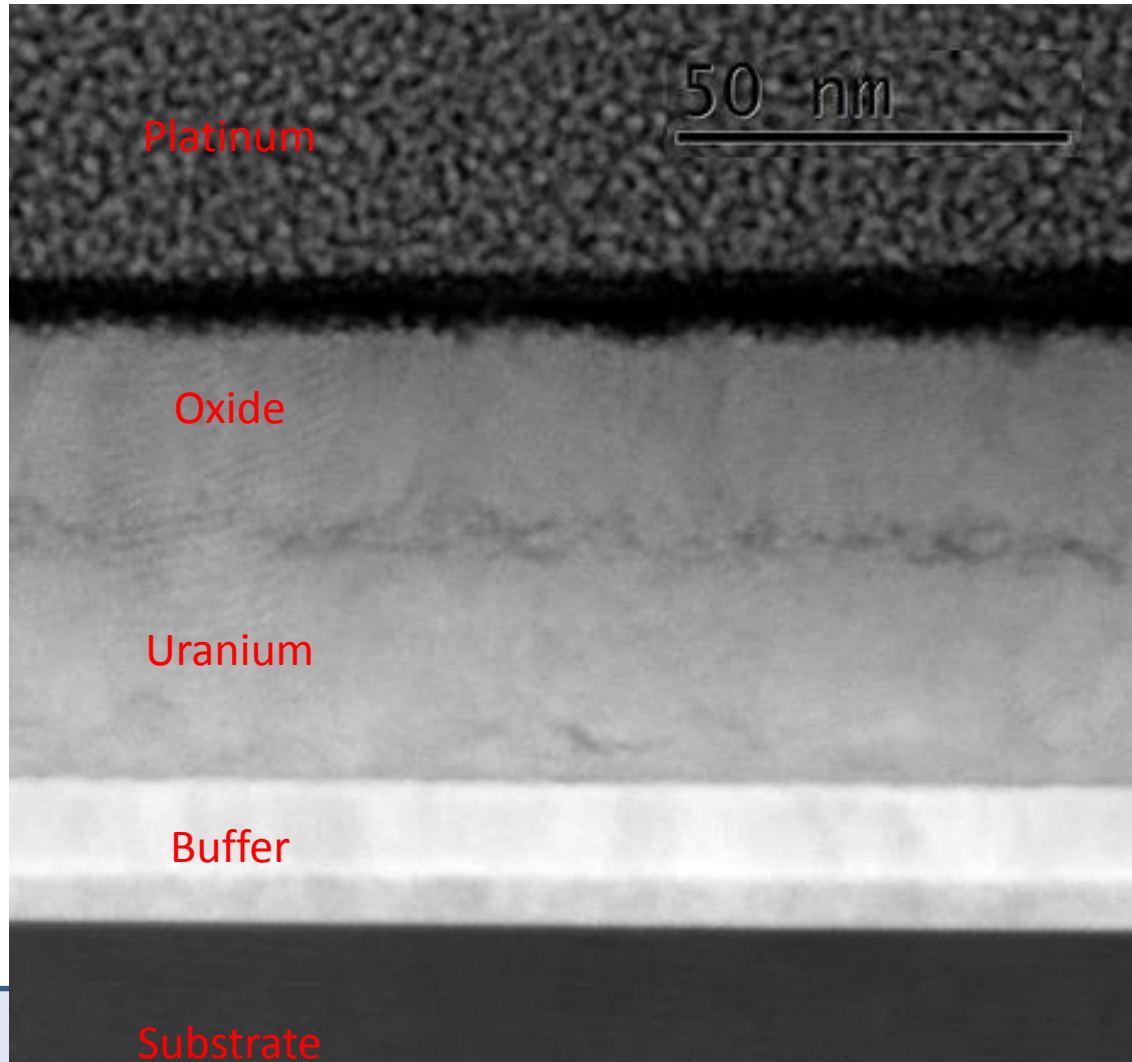
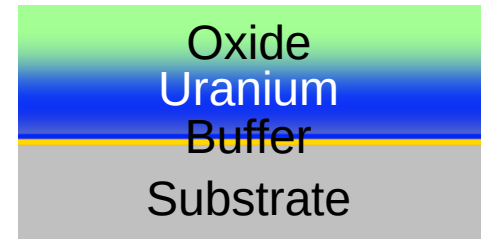
Reflectivity



X-ray Findings

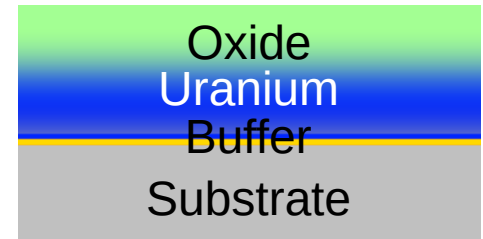
- No identification of crystalline β - UH_3 (seen in bulk U experiments under similar conditions)
- Consumption of U metal (2 mbar P_{H_2} , 80-140°C) monitored in-situ by synchrotron XRD
- Preference of consumption of (110) over (002) demonstrated
- Oxide consumed and strained throughout with higher oxides (UO_{2+x}) also reduced
- Reflectivity shows reduction in U metal density material
- Increase in amount of UO_2 (10.97g/cm³) or UH_3 (10.95 g/cm³) density material
- Roughening of the U/ UO_2 interface

STEM Bright Field

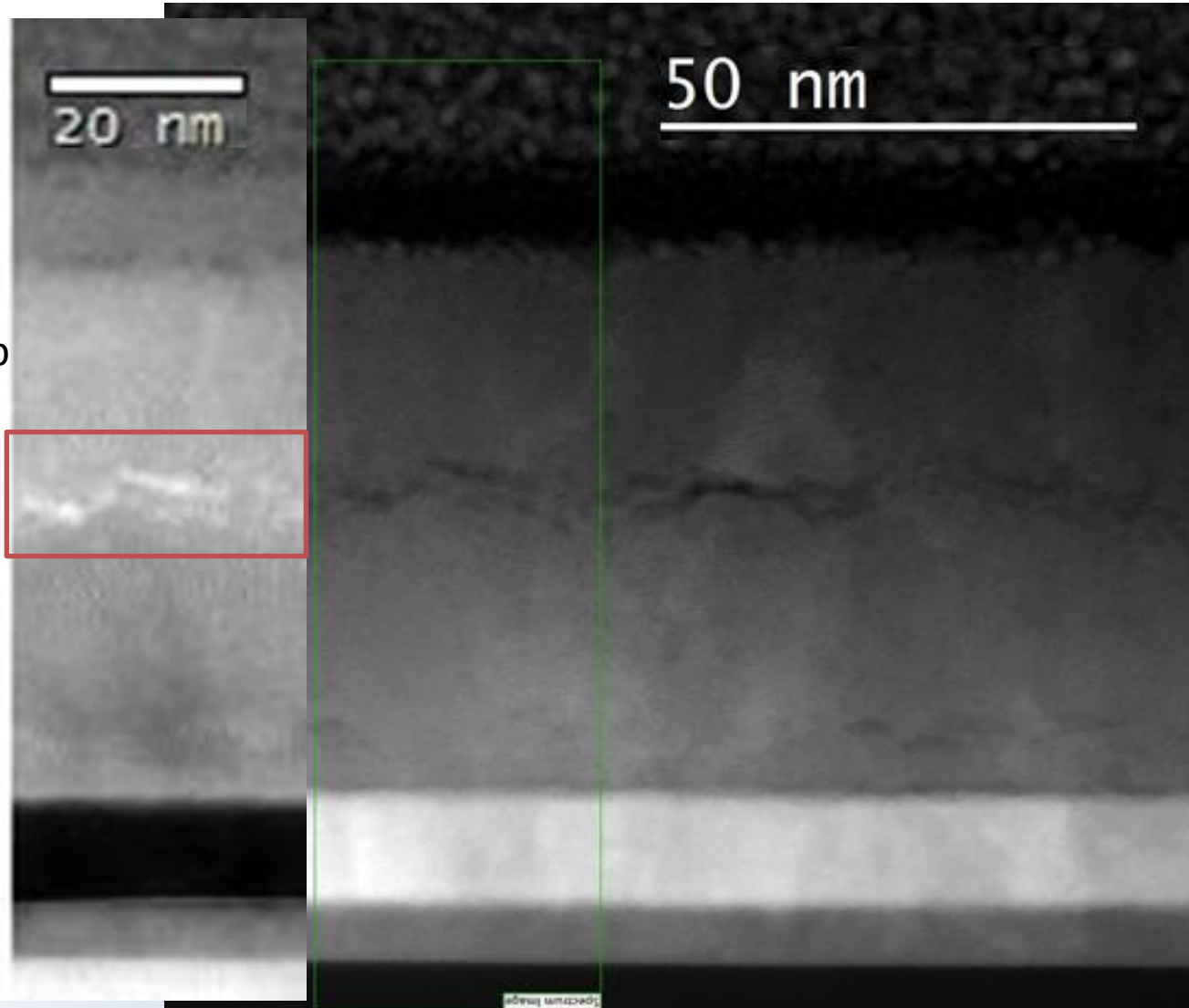


STEM

High Annular Dark Field



EELS
Selected
Peak map



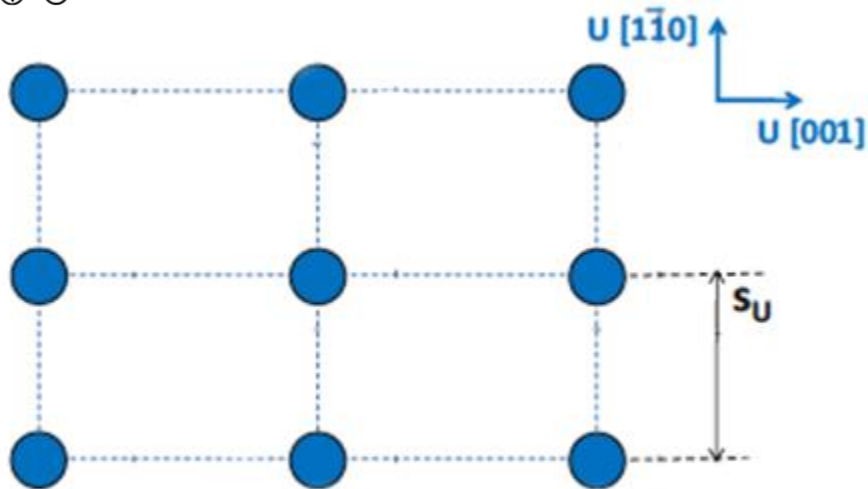
Hydriding of Uranium

- Small partial pressure of H_2 and moderate temperatures over hours forms hydride
- Different Uranium crystallographic directions are preferentially consumed
- Amorphous hydride produced at the interface between U and UO_2
- No crystalline hydride observed

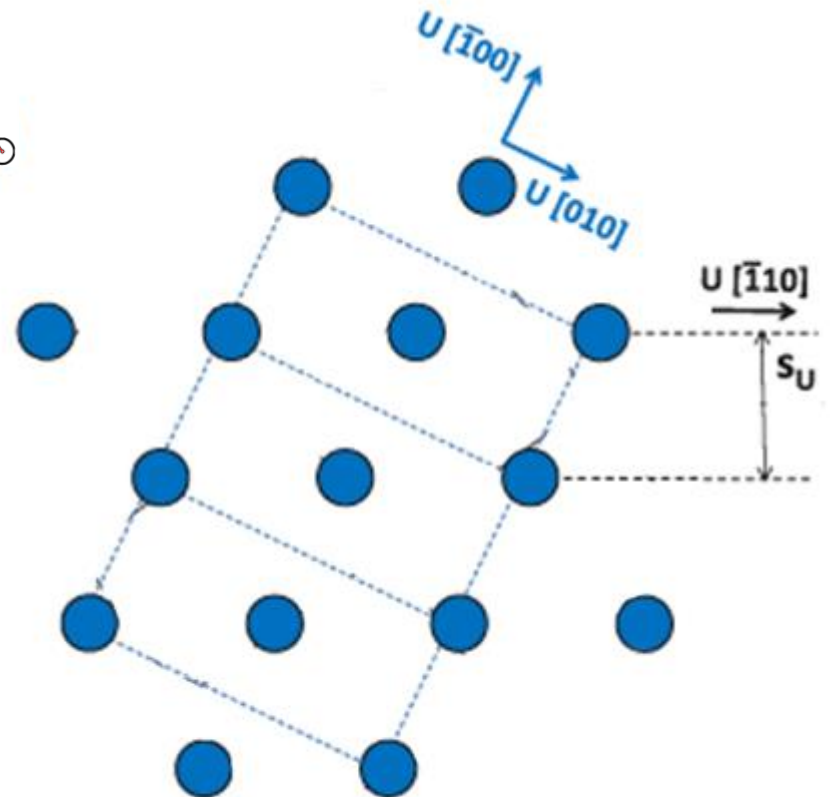
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Crystallographic Dependence



110



002